Routine Production of Cu-61 and Cu-64 at the University of Wisconsin

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The application of copper isotopes in PET research has undergone a dramatic rise, driven by their versatile chelation chemistry, favourable decay characteristics, and national distribution potential. The (p,n) reaction has long been used to produce ⁶¹Cu and ⁶⁴Cu from ⁶¹Ni and ⁶⁴Ni with reported yields of 21.4 \pm 2.2 mCi/uA/hr and 8.7 \pm 0.4 mCi/uA/hr at 11 MeV, respectively.¹ The ⁶⁴Ni(p,n)⁶⁴Cu reaction in particular necessitates careful consideration of incident particle energy. Electrodeposition of enriched ⁶¹Ni and ⁶⁴Ni target material onto high purity gold or silver blanks has been described previously and appears to be limited to approximately 80-120 mg/cm², by time and cost concerns.

Using the pooled cross section data $\sigma(E)$ for the ⁶⁴Ni(p,n)⁶⁴Cu reaction,² the end of saturated (EoSB) yield of ⁶⁴Cu can be predicted as a function of ⁶⁴Ni thickness and incident beam energy, shown below. This family of yield curves strongly suggests that very thick targets ($\approx \frac{1}{2}$ gram/cm²; \approx \$10,000 in ⁶⁴Ni inventory) are needed to take advantage of proton energies above 11 MeV, being prohibitive both in cost and plating time. We have degraded the 16 MeV incident proton energy of the PETtrace to approximately 12 MeV with a 0.23 mm tantalum foil to improve the efficiency of our production runs. However, it is apparent that our legacy CTI RDS 112 is still far better suited for the weekly production of ⁶⁴Cu at the 0.5 Ci level for our own needs, as well as national distribution of the excess.



Copper-61 offers several advantages over ⁶⁴Cu for PET imaging, namely 61% vs 20% β^+ branching and a 3.4 hr vs 12.7 hr half-life, which combine to result in a three-fold greater useful β^+ flux to absorbed radiation dose ratio for trapped agents. Three reactions present themselves for cyclotron facilities without alpha beams: ⁶¹Ni(p,n)⁶¹Cu, ⁶⁰Ni(d,n)⁶¹Cu, and ⁶⁴Zn(p,\alpha)⁶¹Cu. With the

recent three-fold price increase of enriched ⁶¹Ni, we have reverted to the ⁶⁰Ni(d,n)⁶¹Cu reaction for protocols needing Cu-ATSM for hypoxia imaging in human and veterinary patients.³ Human studies use enriched ⁶⁰Ni plated on gold discs. Animal studies, with more relaxed specific activity requirements (>300 mCi/µmole), can utilize the deuteron irradiation of ^{nat}Ni targets, obviating the need for recycling of enriched target stock. The HPGe spectrum below testifies to the radionuclidic purity of the ⁶¹Cu. Electroplated and foil targets are dissolved in HCl at 100^s C, accelerated with H₂O₂. Alternatively, biasing the Ni foil (10 volts, 1 amp) in unheated concentrated HCl removes approximately 40 mg of the foil and >90% of the activity in 3 minutes.⁴ The dissolution apparatus is identical to the electroplating setup. These platers have been recently improved, adding flow, temperature control, pulsed voltage and current regulation under LabView control.

As more subtle targeting strategies develop, the chelation of copper radionuclides to molecular imaging candidates will permit PET to determine the best lead compound, significantly shortening the time to achieve diagnostic utility. Any improvements in the supply of ⁶¹Cu and ⁶⁴Cu will greatly serve that end.



¹ Avila-Rodriguez M A (2007). Low energy cyclotron production of multivalent transition metals for PET imaging and therapy. Ph.D. Dissertation University of Wisconsin Press, Madison, WI.

³ Tolmachev V, Lundqvist H, Einarsson L (1998). Production of 61Cu from a natural nickle target. Applied Radiation Isotopes, 49(1-2), 79-81.

⁴ Martin C C, Oakes T R, Nickles R J (1990). Small Cyclotron Production of Cu-60 PTSM for PET Blood Flow Measurements. J Nucl Med 31, p815.

² Cyclotron Produced Radionuclides: Physical Characteristics and Production Methods (2009). IAEA Technical Reports Series No. 468. IAEA Press, Vienna, Austria.